Organic Molecules and Meteorites

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Organics in Meteorites: Possible sources

Astrophysical / Interstellar formation

Protosolar Nebula

Parent body processes

Decreasing abundances with increasing C number indicate abiotic origin. (cf. Pizzarello et al. 2006, *Meteorites II*)

Basic Meteorite Classification

Irons and Mesosiderites

Dominated by metallic Fe-Ni Formed in cores of differentiated parent body

Achondrites

Formed from crustal material of differentiated parent body Heavily processed

Chondrites

Primitive material from undifferentiated parent body Composed of refractory inclusions, chondrules (glassy melt droplets) and intergranular matrix

Carbonaceous Chondrites and Ordinary Chondrites

Carbonaceous Chondrites

Probably formed further out in solar nebula. Tend to be aqueously altered to unaltered. CI, CM, CR, CV, CO, CK, CH

Ordinary Chondrites

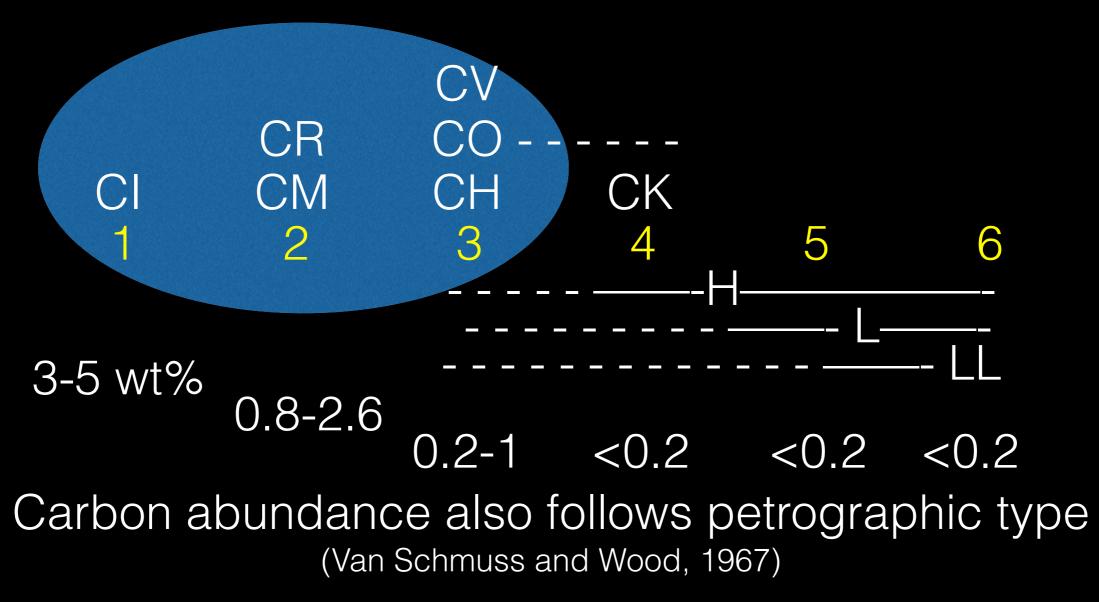
Probably formed closer in to Sun. Tend to be thermally altered. H, L, LL

Enstatite Chondrites

N.B. Carbonaceous Chondrites are not all carbon-rich. Some have virtually no carbon and some OCs have comparable C quantities.

Petrographic Type

Type 3: Little to no alteration Type 2 and 1: Predominantly aqueously altered Types 4, 5, 6: Predominantly thermally altered



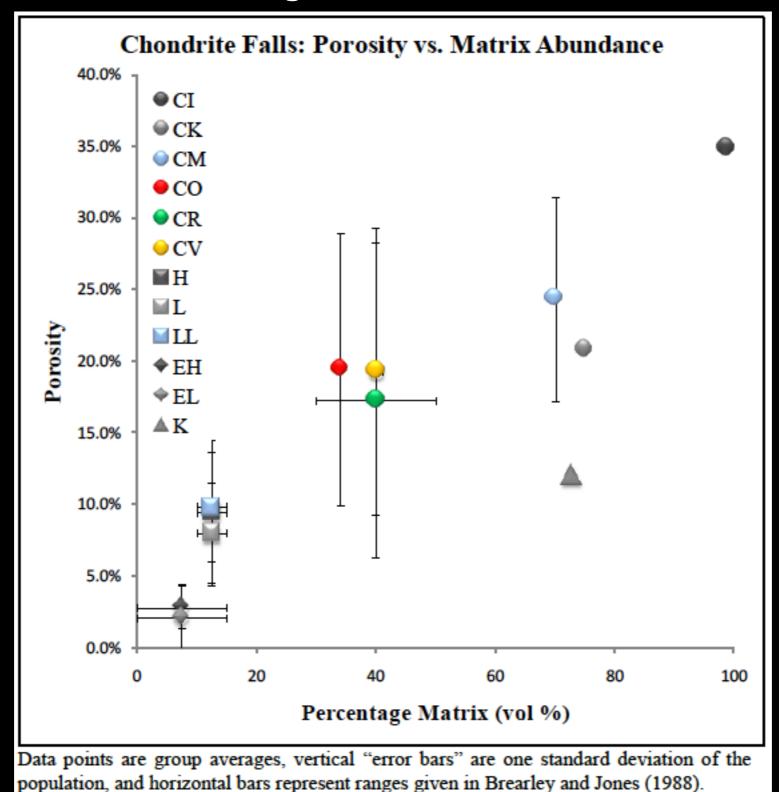
Organics in Meteorites

Predominantly in Carbonaceous Chondrites Why? Low alteration or aqueous alteration

Also:

High temperatures destroy organics such as amino acids. Chondrules and refractory inclusions formed at high temperatures (>1000 K). Matrix remained at low temperature.

Porosity and Matrix



From Macke (2010)

Table 1. The abundances of insoluble and soluble organic compounds (in $\mu g/g$ or ppm) found in carbonaceous chondrites. For the CMs, all data are from the Murchison CM2 meteorite, unless otherwise noted (updated from Botta & Bada 2002).

	(CI		СМ		CR	Tag. Lake
Matrix (vol%)	1	00		~ 50		~ 35	~ 80
IOM	~20	0,00	0 ~	~10,000	-	$\sim 5,000$	~18,000
Amino acids	^	-5 ^a		14-71 ^d	1	$1-250^{b}$	0.04-5.6 ^c
Aromatic hydrocarbons				3^d		16^e	
Aliphatic hydrocarbons				>35			
Monocarboxylic acids				>300		96 ^e	165-448 ^c
Hydroxy- and dicarboxylic acid	ds			14-15		212^{e}	
Purines and pyrimidines				1.3			
Basic N heterocycles				7			
Amines				8		103 ^e	
Alcohols				11			
Aldehydes and Ketones				27			
Sulphonic acids				68			
Phosphonic acids				2			
Polyols				$> 8^{f}$			

Notes: ^a Average for Orgueil and Ivuna (Ehrenfreund et al. 2001).

^bRange from Martins et al. (2007).

^c Herd et al. (2011).

^d For Y-791198 (Naraoka et al. 1998).

^e For GRA 95229 (Pizzarello et al. 2008).

^f Lower limit of glyceric acid (Cooper et al. 2001).

Source: C. M. O'D Alexander (2011) Proc. IAU Symp. 280

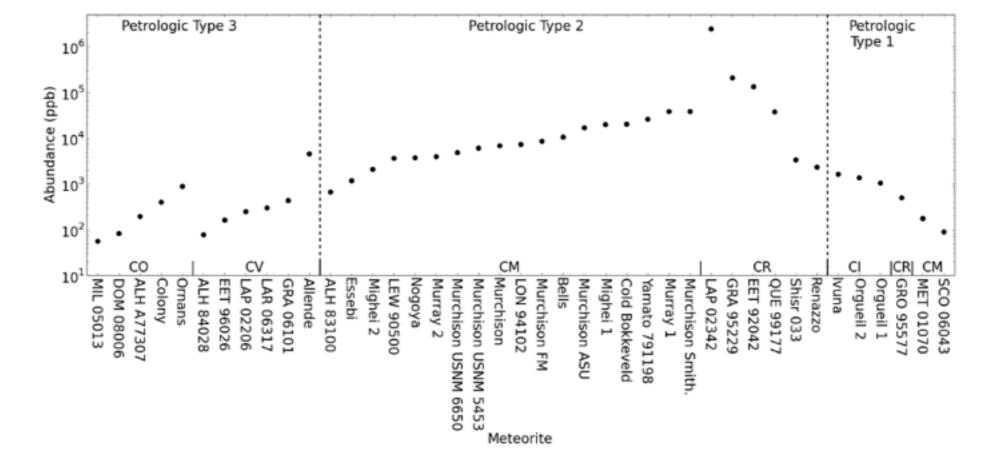
Amino Acids

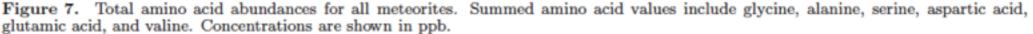
More than 80 amino acid species, both D and L enantiomers in near-racemic abundances. (Terrestrial biological organics have 20, all L)

Slight L-enantiomeric excesses of some amino acids.

Some may be the result of terrestrial contamination. Some may be the result of circularly-polarized UV radiation in interstellar or protosolar ices. (cf. Marcellus et al. 2011, ApJ Lett. 727:L27)

Amino Acids





Source: Cobb and Pudritz (2014) Ap.J. 783

Summary

- Matrix of Carbonaceous Chondrites is the best source for soluble organic material in meteorites.
- While insoluble organic molecules follow total C abundances, amino acids are more abundant in mildly-aqueously-altered (type 2; CR, CM) meteorites.
- Meteorites exhibit a wider variety of organic molecules and isomers than are found in terrestrial sources.
- Organics in meteorites exhibit both D and L enantiomers in near-equal quantities. There is a slight L excess in some amino acid species.
- All of this is consistent with abiotic processes.